Indium and silver recovery from perovskite thin film solar cell waste by means of nanofiltration

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The rapid deployment of perovskite solar cells (PSCs) poses challenges for sustainable raw material supply and waste management of the future, particularly due to their use of precious raw materials such as indium (In) and silver (Ag) [1]. This study presents a novel and environmentally benign process for the recovery of PbI₂, In and Ag from end-of-life PSCs, integrating hydrometallurgical extraction and advanced nanofiltration (NF) technologies.

Following mechanical dismantling and aqueous PbI₂ recovery [2], In and Ag were extracted using nitric acid, achieving recovery efficiencies of $87 \pm 7\%$ for both metals [3]. To separate and concentrate these elements from the acidic leachate, we employed a custom-built layer-by-layer (LbL) NF membrane. The LbL membrane demonstrated near-complete In retention (96.9 \pm 0.4%) and negligible Ag retention, resulting in a concentrated In stream and a dilute Ag permeate. Compared to a commercial acid-resistant NF membrane, the LbL membrane achieved similar separation performance with significantly lower pressure (5 bar vs. 25 bar) and higher flux (30 L/m²h vs. 6 L/m²h), reducing energy consumption by over 85%. By modifying their composition, the LbL membranes gained stability towards a number of highly concentrated inorganic acids, expanding their range of applications beyond In / Ag recovery from PSC [4].

Life cycle assessment confirmed the environmental benefits of the processes in comparison to primary raw material extraction, especially when acid reuse strategies were considered. Hence, the hydrometallurgical recovery of Pb, In and Ag offers a scalable and environmentally benign solution for material recycling from next-generation photovoltaics [3,4].

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^[1] L. Wagner, J. Suo, B. Yang, D. Bogachuk, E. Gervais, R. Pietzcker, A. Gassmann, J.C. Goldschmidt, The Resource Demand of Terawatt-Scale Perovskite Tandem Photovoltaics, (2023). https://doi.org/10.2139/ssrn.4493241.

^[2] F. Schmidt, M. Amrein, S. Hedwig, M. Kober-Czerny, A. Paracchino, V. Holappa, R. Suhonen, A. Schäffer, E.C. Constable, H.J. Snaith, M. Lenz, Organic solvent free PbI2 recycling from perovskite solar cells using hot water, Journal of Hazardous Materials 447 (2023) 130829. https://doi.org/10.1016/j.jhazmat.2023.130829.

^[3] M. Amrein, K. Rohrer, D. Hengevoss, H. Jin, H.J. Snaith, M. Thomann, F. Nüesch, M. Lenz, Indium and silver recovery from perovskite thin film solar cell waste by means of nanofiltration, (2025).

^[4] M. Amrein, K. Rohrer, D. Hengevoss, T. Müller, B. Vallat, D. Rocco, M. Thomann, F. Nüesch, S. Hedwig, M. Lenz, Stability Assessment of Layer-by-Layer Nanofiltration Membranes for Element Recovery from Highly Acidic Media, (2025). https://doi.org/10.2139/ssrn.5208726.